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TITLE OF THE INVENTION (280 characters max)				
POWER CABLE COMPOSITIONS FOR STRIPPABLE ADHESION				
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PROVISIONAL APPLICATION FILING ONLY

POWER CABLE COMPOSITIONS FOR STRIPPABLE ADHESION**FIELD OF THE INVENTION**

This invention relates to power cable compositions. Specifically, it relates to semiconductive power cable compositions, articles (such as a semiconductive cable layer and power cable constructions) prepared from the semiconductive compositions, and
5 processes for preparing the semiconductive compositions and related articles.

DESCRIPTION OF THE PRIOR ART

Power cables, rated for a conductor operating temperature of 90-degree Centigrade or higher, are commonly prepared by extruding chemically-crosslinkable polymer materials around the conductor. Following extrusion, the chemically-
10 crosslinkable polymeric materials are crosslinked to resist material deformation at the rated cable operating temperature and related overload conditions.

For medium- and high-voltage cable designs, the chemically-crosslinkable polymeric materials commonly contain conductive fillers to render the resulting cable layer semiconductive. The chemically-crosslinkable polymeric materials are extruded to
15 prepare an electrical stress control layer between the metallic conductor and a polymer-dielectric insulation layer and may also be used as an electrical stress control layer between the polymer dielectric layer and grounding wires or tapes. The various layers are typically co-extruded and subsequently, simultaneously crosslinked.

Co-extrusion and simultaneous crosslinking are generally desirable because the
20 resulting cable layers are closely bonded. Close bonding prevents partial delamination of the layers and precludes void forming between the layers, thereby preventing premature

cable failure. Delamination and void formation can result from flexure and/or heat during the normal use of the cable.

Unfortunately, close bonding as a result of co-extrusion and simultaneous crosslinking is not free of disadvantages. Notably, the method of manufacture presents
5 problems for applications in which stripping the outermost electrical stress control layer (or semiconductive layer) from the polymer-dielectric insulation layer is desirable. (It is believed that crosslinking bond occur across the interface between the electrical stress control layer and the polymer-dielectric insulation layer. Those bonds must be broken to strip the layers apart.) Stripping the semiconductive layer away from the insulation layer
10 results in damaging the insulation layer.

It is desirable for the semiconductive layer to adhere to the insulation layer under normal operating conditions while being easily strippable from the insulation layer on demand. These features promote utilization of the cable for its normal life and ease of installation of such accessories as joints, splices, and terminations.

15 Currently, the chemically-crosslinkable polymeric materials often contain polar polymers to reduce their melt miscibility with insulation materials, which generally contain non-polar polyolefinic polymers. Most commonly, the chemically-crosslinkable polymer materials are based upon ethylene vinyl acetate copolymers, having a vinyl acetate comonomer content of greater than 28% by weight. A disadvantage of these
20 high-polarity copolymers is that they tend to yield compounds prone to agglomeration. It is desirable to avoid the problem of agglomeration.

SUMMARY OF THE INVENTION

The present invention is a semiconductive power cable composition comprising (a) a mixture of a high-temperature polymer and a soft polymer, and (b) a conductive filler, wherein a semiconductive cable layer prepared from the composition strippably adheres to a second cable layer. The invention also includes a semiconductive cable layer prepared from the semiconductive power cable composition as well as a power cable construction prepared by applying the semiconductive cable layer over a wire or cable.

Moreover, the present invention includes a process for preparing the semiconductive power cable composition comprising the step of blending a mixture of a high-temperature polymer, a soft polymer, and a conductive filler. Alternatively, the process comprises the steps of (a) reactively-coupling a mixture of a high-temperature polymer, a soft polymer, and a first coupling agent, in the presence of a conductive filler and wherein the resulting mixture having a reduced curative level, and (b) admixing a second coupling agent, wherein the second coupling agent does not substantially affect the curative level of the resulting mixture.

The present invention also includes a process for preparing a power cable comprising the steps of (a) extruding a power cable semiconductive composition over a metallic conductor to yield a semiconductive cable layer over the metallic conductor, (b) extruding a chemically-crosslinkable insulation composition over the semiconductive cable layer, (c) extruding a second semiconductive power cable composition over the polymer-dielectric insulation to yield a second semiconductive cable layer, and (d)

crosslinking the chemically-crosslinkable insulation composition to yield a crosslinked, polymer-dielectric insulation.

DESCRIPTION OF THE INVENTION

The invented semiconductive power cable composition comprises (a) a mixture of
5 a high-temperature polymer and a soft polymer and (b) a conductive filler, wherein a
semiconductive cable layer prepared from the composition strippably adheres to a second
cable layer. As the term is used herein, "strippably adheres" means that the
semiconductive layer adheres to a second layer (usually, an insulation layer) under
normal operating conditions of the power cable while having the property of being easily
10 strippable from the second layer (i.e., delaminating/separating the semiconductive layer
from the second layer without substantially damaging the second layer) on demand.

A high-temperature polymer, as that term is used herein, means a polymer having
suitable heat resistance for the semiconductive cable layer but lacking other desirable
properties. For example, the high-temperature polymer may not have desirable
15 processing characteristics or other material properties. Suitable high-temperature
polymers include polypropylenes, polyesters, nylons, polysulfones, and polyaramides.
Preferred high-temperature polymers are polypropylenes.

A soft polymer, as that term is used herein, means a polymer that enhances the
processing characteristics of the high-temperature polymer and provides a networking
20 source as the soft polymer is coupled to the high-temperature polymer for additional heat
resistance. Suitable soft polymers include polyethylenes, polypropylenes, polyesters, and
rubbers. Preferred soft polymers are polyethylenes.

Polyethylenes include homopolymers of ethylene and copolymers of ethylene and one or more alpha-olefins, and, optionally, a diene. The polyethylene can also be a copolymer of ethylene and an unsaturated ester such as a vinyl ester (e.g., vinyl acetate or an acrylic or methacrylic acid ester), a copolymer of ethylene and an unsaturated acid
5 such as acrylic or methacrylic acid, or a copolymer of ethylene and a vinyl silane (e.g., vinyltrimethoxysilane and vinyltriethoxysilane) as well as interpolymers of any of these comonomers. Post-modified polyethylenes of any other of the above are considered within the scope of this invention as well as blends thereof. Preferred polyethylenes are homopolymers of ethylene, copolymers of ethylene and one or more alpha-olefins, and a
10 copolymer of ethylene and an unsaturated ester. More preferred polyethylenes for soft polymers are copolymers of a polar monomer and a nonpolar comonomer. Most preferred polyethylenes are copolymers of ethylene and an unsaturated ester.

Suitable polypropylenes include homopolymers of propylene, copolymers of propylene and other olefins, and terpolymers of propylene, ethylene, and dienes.

15 Suitable polyesters include thermoplastic resins comprising a saturated dicarboxylic acid and a saturated difunctional alcohol. Specific examples include polyethylene terephthalate, polypropylene terephthalate (or trimethylene terephthalate), polybutylene terephthalate, polytetramethylene terephthalate, polyhexamethylene terephthalate, polycyclohexane-1,4-dimethylol terephthalate, and polyneopentyl
20 terephthalate. Preferred polyesters are polyethylene terephthalate, polypropylene terephthalate (or trimethylene terephthalate), and polybutylene terephthalate.

Suitable nylons include nylon 6, nylon 6,6, and nylons based upon longer chain-length diamines. Preferred nylons are nylon 6 and nylon 6,6.

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Suitable rubbers include thermoplastic rubbers, ethylene propylene diene rubber, styrene-butadiene block copolymers, styrene-butadiene rubber, polybutadiene rubbers, isoprene rubbers, nitrile rubbers, polychloroprene rubbers, hydrogenated styrene-butadiene block copolymers, methacrylate butadiene styrene rubber, acrylic elastomers
5 (such ethylene methylacrylate), fluoroelastomers, and thermoplastic elastomers (such as thermoplastic urethanes, polyamids, and polyester ethers).

Suitable conductive fillers include carbon blacks, carbon fibers, carbon nanotubes, graphite particles, metals, and metal-coated particles. Preferred conductive fillers are carbon blacks.

10 In addition, the semiconductive power cable composition may further comprise a coupling agent. The term "coupling agent," as used herein, means a compound or mixture of compounds used for the purposes of coupling or grafting a polymer or polymer blend. The coupling agent may be present in an amount sufficient to reduce the curative level of the mixture of the high-temperature polymer and the soft polymer.

15 An amount of a second coupling agent may also be present in the semiconductive composition. The second coupling agent may be the same compound as the first coupling agent. The second coupling agent does not substantially affect the curative level of the mixture of the high-temperature polymer and the soft polymer.

Suitable coupling agents include organic peroxides, azides, organofunctional
20 silanes, maleated polyolefins, phenols, and sulfur vulcanizing agents. Suitable organic peroxides include aromatic diacyl peroxides, aliphatic diacyl peroxides, dibasic acid peroxides, ketone peroxides, alkyl peroxyesters, and alkyl hydroperoxides. Suitable azide coupling agents include alkyl azide, aryl azides, acyl azides, azidoformates,

phosphoryl azides, phosphinic azides, silyl azides, and polyfunctional azides. Suitable silanes include unsaturated silanes that comprise an ethylenically unsaturated hydrocarbyl group, such as a vinyl, allyl, isopropenyl, butenyl, cyclohexenyl or γ -(meth)acryloxy allyl group, and a hydrolyzable group, such as, for example, a hydrocarbyloxy, hydrocarbonyloxy, or hydrocarbylamino group. Preferred coupling agents are organic peroxides.

In addition, the semiconductive power cable composition may further comprise a compatibilizing polymer. As used herein, the term "compatibilizing polymers" includes those polymers having an affinity for both the high-temperature polymer and the soft polymer. Preferred compatibilizing polymer are copolymers (such as ethylene-alpha-olefin copolymers) and functionalized polymers (such as maleated polyolefins and glycidil-functional polyolefins). Based the high-temperature and soft polymers, a person skilled in the art can readily identify other suitable compatibilizing polymers.

In addition, the composition may contain other additives such as antioxidants, stabilizers, blowing agents, pigments, processing aids, and cure boosters.

In a preferred embodiment, the present invention is a semiconductive power cable composition comprising (a) a mixture of a high-temperature polymer and a soft polymer and (b) a conductive filler, wherein a semiconductive cable layer prepared from the composition strippably adheres to a second cable layer. The high-temperature polymer and the soft polymer may have different heat resistance. In a more preferred embodiment, the second cable layer is a chemically-crosslinked layer.

In an alternate embodiment, a semiconductive cable layer is prepared from the semiconductive power cable composition. In a yet another embodiment, a power cable construction is prepared by applying the semiconductive cable layer over a wire or cable.

5 In another alternate embodiment, the present invention is a process for preparing a semiconductive power cable composition comprising the step of blending a mixture of a high-temperature polymer, a soft polymer, and a conductive filler, wherein a semiconductive cable layer prepared from the composition strippably adheres to a second cable layer. In this embodiment, the mixture may further comprise a coupling agent in an amount to reduce the curative level of the mixture.

10 In yet another embodiment, the invention is a process for preparing a semiconductive power cable composition comprising the steps of (a) reactively-coupling a mixture of a high-temperature polymer, a soft polymer, and a first coupling agent, in the presence of a conductive filler and wherein the resulting mixture having a reduced curative level, and (b) admixing a second coupling agent, wherein the second coupling
15 agent does not substantially affect the curative level of the resulting mixture, wherein a semiconductive cable layer prepared from the composition strippably adheres to a second cable layer.

In another embodiment of the present invention, the invention is a process for preparing a power cable comprising the steps of (a) extruding a semiconductive power
20 cable composition over a metallic conductor to yield a semiconductive cable layer over the metallic conductor, and (b) extruding a polymer-dielectric insulation over the semiconductive cable layer. This embodiment may further comprise the step of (c)

extruding a second semiconductive power cable composition over the polymer-dielectric insulation to yield a second semiconductive cable layer.

In an alternate aspect of this embodiment, the invention is a process comprising the steps of (a) extruding a power cable semiconductive composition over a metallic
5 conductor to yield a semiconductive cable layer over the metallic conductor, (b) extruding a chemically-crosslinkable insulation composition over the semiconductive cable layer, (c) extruding a second semiconductive power cable composition over the polymer-dielectric insulation to yield a second semiconductive cable layer, and (d)
crosslinking the chemically-crosslinkable insulation composition to yield a crosslinked,
10 polymer-dielectric insulation.

EXAMPLES

The following non-limiting examples illustrate the invention.

Examples 1-7

In Comparative Examples 1, 6, and 7 and Examples 2; 4, and 5, the mixtures were
15 combined in a lab-scale compounder to achieve a melt temperature of 190 degrees Centigrade for 5 minutes. In Example 3, the mixture of the high-temperature polymer, the soft polymer, and the conductive filler were combined in a lab-scale compounder to achieve a melt temperature of 190 degrees Centigrade for 5 minutes and then allowed to cool; then the peroxide was added at 120 degrees Centigrade.

20 Each exemplified formulation was evaluated for Hot Creep performance and adhesion to a polyethylene-insulation substrate. The Hot Creep test specimens were evaluated for their resistance to thermal deformation under load conditions of 20 N/sq.

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cm. tensile stress for 15 minutes at 150 degrees Centigrade. Elongation and residual deformation were measured. Residual deformation is reported in Table I as % Hot Set.

For the adhesion measurement, 30 mil plaques of the exemplified formulations were prepared. A polyethylene-insulation substrate was prepared from The Dow Chemical Company's commercially available HFDB-4202 crosslinkable polyethylene insulation at 120 degrees Centigrade. Subsequently, the test plaques and the polyethylene-insulation substrate were molded together under pressure at a temperature in excess of 180 degrees Centigrade for a length time sufficient for the substrate to cure. Next, the dual-layer specimens were conditioned at ambient temperature overnight. A one-half inch wide strip was scored from the dual-layer specimen. A 90-degree peel test was performed in an INSTRON™ tensile machine at a peel rate of 20 inches per minute.

The polymeric materials for the exemplified formulations were added in the concentrations shown in Table I and include:

- (1) duPont Elvax 265™ ethylene vinylacetate copolymer (EVA-1), having a vinylacetate content of 28% by weight and a melt index of 3 g/10minutes;
- (2) DXM-451™ ethylene vinylacetate copolymer (EVA-2), having a vinylacetate content of 18% by weight and a melt index of 3 g/10minutes and commercially available from The Dow Chemical Company;
- (3) 5D45™ polypropylene (PP-1), which was a fractional homopolymer having a melt flow rate of 0.8 and commercially available from The Dow Chemical Company; and
- (4) a homopolymer of polypropylene (PP-2) having a melt flow rate of 20.

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The carbon black was CSX614 and commercially available from Cabot Corporation. Each formulation contained 55 parts per hundred polymer (pphr) of carbon black. The peroxide used was TRIGANOX 101, and commercially available from Akzo Nobel. For Comparative Example 1 and Examples 2-5, the peroxide was added in the amount of 0.4 pphr during the compounding. For Comparative Example 6, the peroxide was added in the amount of 0.4 pphr following the compounding. The formulation for Comparative Example 7 did not contain any peroxide.

While the peroxide in the formulation exemplified as Comparative Example 1 was fully reacted during compounding, the test specimens were unable to withstand the Hot Creep load. Without any residual unreacted peroxide, the test specimen for Comparative Example 1 was fully bonded to the polyethylene-insulation substrate. This bonding also demonstrates that, if the formulation had not contained any peroxide during the compounding, the resulting test specimen would have fully bond itself to the polyethylene-insulation substrate and that current practices using EVA-2 would result in a fully-bonded test specimen.

Table I shows that formulation exemplified as Example 2 had desirable Hot Creep and strippability from the polyethylene-insulation substrate. While the Example 3 formulation showed improved heat resistance, its test specimens became fully bonded to the polyethylene-insulation substrate. Examples 4 and 5 demonstrated formulations with improvements over the Example 2 formulation in heat resistance and strippability.

While Comparative Example 6 demonstrated excellent heat resistance, its unreacted peroxide (i.e., peroxide not consumed during compounding) did not sufficiently reduce the curative level of the mixture to prevent the test specimens from

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fully bonding to the polyethylene-insulation substrate. Comparative Example 7, which is a blend with no coupling agent, yielded test specimens with relatively poor heat resistance.

TABLE I

Component	Comp. Ex. 1	Ex. 2	Ex. 3	Ex. 4	Ex. 5	Comp. Ex. 6	Comp. Ex. 7
EVA-1				65	65		
EVA-2	100	65	65			65	65
PP-1		35		35		35	35
PP-2			35		35		
Test							
Adhesion	Bonded	2.5	Bonded	1.1	3.8	Bonded	Bonded
Hot Creep %	Failed	27	18	18	18	0	56
Hot Set %	---	6.3	5.5	1.6	0.8	0.1	22

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ABSTRACT

The present invention is a semiconductive power cable composition made from or containing (a) a mixture of a high-temperature polymer and a soft polymer, and (b) a conductive filler, wherein a semiconductive cable layer prepared from the composition strippably adheres to a second cable layer. The invention also includes a semiconductive
5 cable layer prepared from the semiconductive power cable composition as well as a power cable construction prepared by applying the semiconductive cable layer over a wire or cable.